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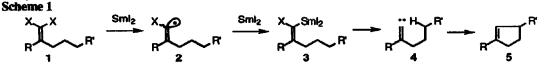
Alkylidenecarbenes from 1,1-Dihalogenoalkenes with Samarium Diiodide: Mild and Efficient Method for the Synthesis of Cyclopentenes

Munetaka Kunishima, Kazuhito Hioki, Shohei Tani,* and Akira Kato[†]

Faculty of Pharmaceutical Sciences, Kobe Gakuin University, Nishi-ku, Kobe 651-21, Japan [†]Nügata College of Pharmacy, Kamishin'ei-cho, Nügata 950-21, Japan

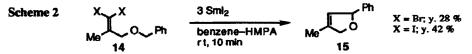
Abstract: A new mild method for construction of cyclopentenes via 1,5-C-H insertion of alkylidenecarbenes by the reaction of 2,2-dialkyl-1,1-dihalogencalkenes with Sml2 in benzene-hexamethylphosphoric triamide (HMPA) was developed.

The intramolecular 1,5-C—H insertion of alkylidenecarbenes is an efficient and convenient method for construction of five-membered ring because the carbenes are capable of undergoing regioselective carbon—carbon bond formation on non-activated carbon centers. Several mild methods for generation of alkylidenecarbenes; fluoride-induced α -elimination of α -chlorovinylsilanes or silylvinyl triflates,^{1a,b} thermal decomposition of tosylazoalkenes^{1c} or diazoalkenes^{1d} and weak base-induced α -elimination of alkenylidonium salts,^{1e} have been developed. Here we report a new mild method for construction of cyclopentenes (5) via 1,5-C—H insertions of alkylidenecarbenes (4) (or carbenoids (3)) generated from 1,1-dihalogenoalkenes (1) by the reaction with SmI2.²



The reaction will proceed by a route illustrated in Scheme 1. Since alkyl groups at α position to carbenic center are loath to migrate leading to alkynes in alkylidenecarbenes, ^{1d, 1e, 3} we examined the reaction of 2,2-dialkyl-1,1-dibromoalkene with SmI2 in benzene—HMPA.⁴ Exposure of 6 (X = Br) in benzene to SmI2 (2.5 equvi.) in benzene—HMPA (9 : 1) at room temperature under nitrogen flow for 10 min gave cyclopentene (7) in 59 % yield (Table 1). Diiodide (6, X = I) also afforded 7 in 76 % yield whereas the reactivity of dichloride toward SmI2 was very low. Another product of these reactions was fully reduced alkene (6, X = H) and no rearranged alkynes were detected. As summarized in Table 1, diiodides generally gave the corresponding insertion product more than dibromides. The reactions of dihalides (8) and (10) afforded fused bicyclic (9) and spiro (11) compounds, respectively. Although a heteroatom activates C—H bond at α position to it toward insertion by carbenes,⁵ the reaction of benzyl ether (14) with SmI2 afforded dihydrofuran derivative (15) in somewhat low yield (Scheme 2). Since the activation of C—H bond is ascribed to weakening of it by α -heteroatoms, an intramolecular hydrogen abstraction by an α -halovinyl radical corresponding to 2 from the activated bond probably competes with the generation of carbenoid.⁶

It should be noted that not only mildness and simplicity of the reaction but also stability and accessibility of *gem*-dihalogenoalkenes make the present reaction useful for synthesis of five-membered ring especially substituted cyclopentenes via alkylidenecarbenes.



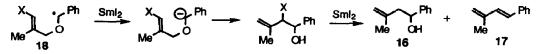
1,1-dihalogenoalkene		cyclization product		yield of reduced
	X	yield (%)*	product (X = H)
X X	CI	3	<i>n</i> -C ₅ H ₁₁	19 ^d
m. I au	CI	15 ^b	m. J	20
₩ ~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	CI	40 ^c		23
6	Br	59	7	10
	I	76		8
XX 8	Br I	56 79	9 Ph	6 3
X X Ph	Br I	51 65		6 5
	Br I	32 52		'n
12			13	

Table 1. 1,5-C-H Insertion of alkylidenecarbenes derived from 1,1-dihalogenoalkenes with SmI2.

a) Reactions were carried out in benzene—HMPA (9:1) for 10 min at r.t. under nitrogen flow. b) The reaction was carried out for 3 h at r.t. c) The reaction was carried out for 3 h at reflux temp. d) 48 % of 6 was recovered.

References and Notes

- (a) Cunico, R. F.; Han, Y. K. J. Organomet. Chem. 1976, 105, C29-C31. (b) Stang, P. J.; Christensen, S. B. J. Org. Chem. 1981, 46, 823-824. (c) Stang, P. J.; Fox, D. P. J. Org. Chem. 1977, 42, 1667-1669. (d) Gilbert, J. C.; Giamalva, D. H.; Weerasooriya, U. J. Org. Chem. 1983, 48, 5251-5256. (e) Ochiai, M.; Takaoka, Y.; Nagao, Y. J. Am. Chem. Soc. 1988, 110, 6565-6566.
- 2 Recently, we have found that the reaction of gem-dibromoalkenes having hydrogen or aryl group at β-position with SmI₂ in benzene—HMPA generates alkylidenecarbenes, which undergo 1,2-rearrangement of an α-substituent to lead alkynes, under mild conditions, see Kunishima, M.; Hioki, K.; Ohara, T.; Tani, S. J. Chem Soc., Chem. Commun. 1992, 219-220.
- 3 Gilbert, J. C.; Weerasooriya, U. J. Org. Chem. 1982, 47, 1837-1845.
- 4 It is important to use benzene to avoid a hydrogen atom abstraction, which occurs predominantly in tetrahydrofuran (THF) to give reduced alkenes, from solvent by an α -halogenovinyl radical (2).²
- 5 Gilbert, J. C.; Blackburn B. K. J. Org. Chem. 1986, 51, 3656-3663, and references cited therein.
- 6 In the reaction of 14 (X = Br), alcohol (16, 28%) and diene (17, 17%) were produced. Formation of these products would be attributed to the intramolecular hydrogen abstraction of α -halovinyl radical from the C—H bond α to the etheral oxygen. The resulting α -oxyradical (18) would be then reduced to the carbanion followed by 2,3-rearrangement, and final reductive dehalogenation afforded 16 and 17.



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